
inducfur Documentation

Release 1.0

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README of Induction Furnace

1.1 Presentation

Some notices and reflections to build an induction furnace.

You can view the docs on [readthedocs](#) and clone the sources from [github](#)

1.2 Getting started

In a bash-terminal:

```
git clone https://github.com/charlyoleg/induction_furnace.git
cd induction_furnace
npm install
npm run install_py
npm run the_docs
```


2.1 Presentation

The *metal* is an interesting material because of its *rigidity*. This property let us make mechanical parts that are closed to an ideal solid, which shapes and dimensions do not vary regardless of the applied strength and temperatures.

The high rigidity of the metal is also an issue when it comes to form the shape of the parts. Direct strength become inefficient. Two methods let us bypass the rigidity property of the metal:

- the electro-erosion takes advantage of the electrical conductivity of the metal
- the fusion (i.e. foundry) takes advantage of the relative small quantity of energy required to melt a metal

For years, the difficulty was to bring the thermal energy to the metal at high temperature. But nowadays, thanks to the induction technology, it is quiet easy to bring this energy as the heat is generated within the metal.

2.2 Induction overview

An LC-circuit generates an alternative magnet field, which generates *Foucault current* (a.k.a. Eddy current), which generates heat thanks to the Ohmic law.

The Ohmic law:

$$W_{th} = R * I^2$$

Wth : the thermal power generated by a current

The Foucault current are limited to the area closed to the surface of the metal if the frequency is too high. This is the *skin effect*

Thickness of the skin effect:

$$Thickness = \frac{1}{\sqrt{conductivity * permeability * \pi * frequency}}$$

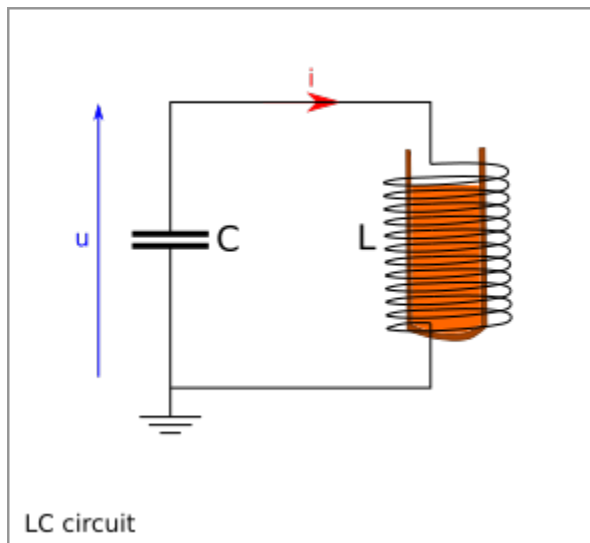
$$frequency = \frac{1}{Thickness^2 * \pi * conductivity * permeability}$$

| Material | Iron | Copper |
|-----------|---------|---------|
| Thickness | | |
| 10 mm | 0.05 Hz | 43 Hz |
| 8 mm | 0.08 Hz | 67 Hz |
| 6 mm | 0.1 Hz | 119 Hz |
| 4 mm | 0.3 Hz | 269 Hz |
| 2 mm | 1.2 Hz | 1075 Hz |
| 1 mm | 5 Hz | 4300 Hz |

To get enough thickness for the *Foucault current*, we work with a frequency below 1kHz. So we can melt all metal. Also, at high temperature, the iron loses its high magnetic permeability, which makes it behave closer to the copper regarding the skin effect.

2.3 LC circuit

To create an alternative magnetic field, we need an oscillating circuit like the LC-circuit.



Formula:

$$u = L \frac{di}{dt}$$

$$i = C \frac{du}{dt}$$

$$f = \frac{1}{2\pi\sqrt{LC}}$$

$$E_{cap} = \frac{Cu^2}{2}$$

$$E_{ind} = \frac{Li^2}{2}$$

2.4 Inductance sizing

The volume of 1kg of iron is 127000 mm³. In chip, it may requires the double volume, that means 250000 mm³ or 50*50*3.14*30, a cylinder of radius 50 mm and height 30 mm.

Cylindrical inductance:

$$L = \mu_0 N^2 \frac{A}{l}$$

$$\mu_0 = 4\pi 10^{-7} Hm^1$$

Considering a cylindrical inductance of radius 80 mm (diameter: 160 mm) and a height of 30 mm. If the wire loop are distance of 5 mm, we get 6 loops for one way. For both ways, we get 12 loops:

$$A = 0.08^2 \pi = 0.02m^2$$

$$l = 0.03m$$

$$N = 12$$

$$L = 0.00012H = 0.12mH$$

2.5 Capacitor sizing

To reach a resonance frequency of 1kHz, we need a capacitor of:

$$C = \frac{1}{(2f\pi)^2 * L}$$

$$C = 0.000211F = 0.211mF$$

2.6 Power transfer

We want to transfer the energy from electricity to heat up to 1kW i.e. 1kJ/s. We have 1000 oscillations per second.

If each oscillation transfers completely its energy in one oscillation, we need 1J in the LC-oscillation:

$$u = \sqrt{\frac{2E_{cap}}{C}} = 97.3(V)$$

$$i = \sqrt{\frac{2E_{ind}}{L}} = 129.1(A)$$

If 10% of the energy is transfer per oscillation, we need 10J oscillating in the LC-circuit:

| |
|--|
| $u = 307 \text{ (V)}$ $i = 408 \text{ (A)}$ |
|--|

If 1% of the energy is transfer per oscillation, we need 100J oscillating in the LC-circuit:

| |
|---|
| $u = 973 \text{ (V)}$ $i = 1291 \text{ (A)}$ |
|---|

CHAPTER 3

Notes on Metal

3.1 Presentation

When looking for a *rigid* substance, metal is a good choice compare to stone, wood, glass, ceramic, plastic. The *rigidity* is relevant when designing parts with the *solid mechanics*. In this matter, shapes and dimensions are considered invariant, regardless of the strength, pressure, collision and temperature conditions.

3.2 Metal overview

3.2.1 Rigidity

| Material | Young's modulus | Yield stress | Yield strain | Thermal expansion |
|-----------------|-----------------|--------------|--------------|-------------------|
| Units | GPa | MPa | % | 10**-6/°C |
| Iron (Fe) | 196 | 80 | 0.04 | 11.8 |
| cast iron | 83 | 400 | 0.4 | 10.5 |
| Steel | 204 | 1450 | 0.7 | 17 |
| Aluminium (Al) | 69 | 95 | 0.13 | 23.1 |
| Aluminium alloy | 75 | 470 | 0.64 | |
| Copper (Cu) | 124 | 33 | 0.03 | 17 |
| Zinc (Zn) | 78 | | | |

Example of thermal expansion:

An iron-part with a width of 10 cm gets longer of 11.8 um when the temperature ↪ increases of 10 °C.

Example of limit of elasticity:

An iron-part under strength **with** a width of 10 cm gets longer of 200 um before ↪ getting a plastic deformation.

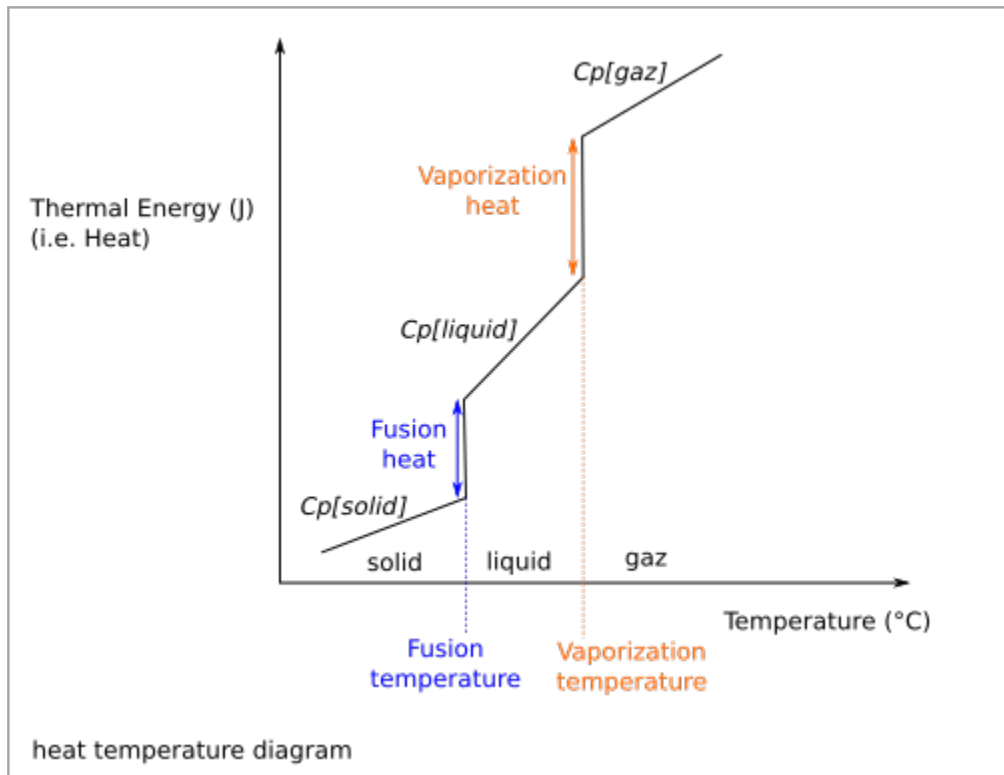
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3.2.2 Auxiliary

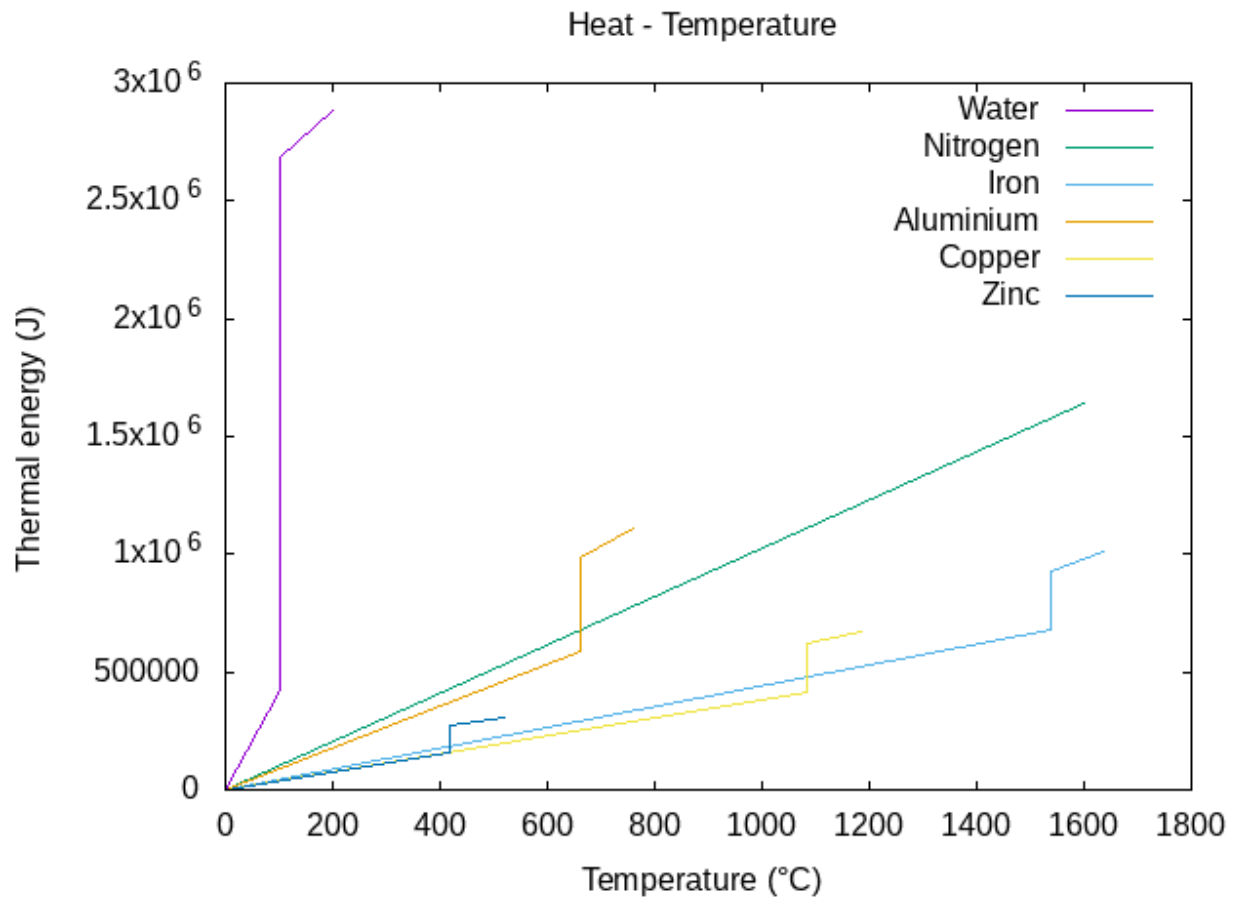
| Material | Molar mass | Specific volume | Specific heat | Thermal conductivity | Electrical resistivity |
|----------------|------------|-----------------------|---------------|----------------------|------------------------|
| Units | kg/mol | m ³ /kg | J/(kg.K) | W/(m.K) | Ohm.m |
| Iron (Fe) | 0.05584 | $127.0 \cdot 10^{-6}$ | 444 | 80.4 | $9.70 \cdot 10^{-8}$ |
| Aluminium (Al) | 0.02698 | $370.4 \cdot 10^{-6}$ | 897 | 237 | $2.65 \cdot 10^{-8}$ |
| Copper (Cu) | 0.06354 | $111.6 \cdot 10^{-6}$ | 385 | 384.1 | $1.68 \cdot 10^{-8}$ |
| Zinc (Zn) | 0.06539 | $140.0 \cdot 10^{-6}$ | 380 | 116 | $5.90 \cdot 10^{-8}$ |

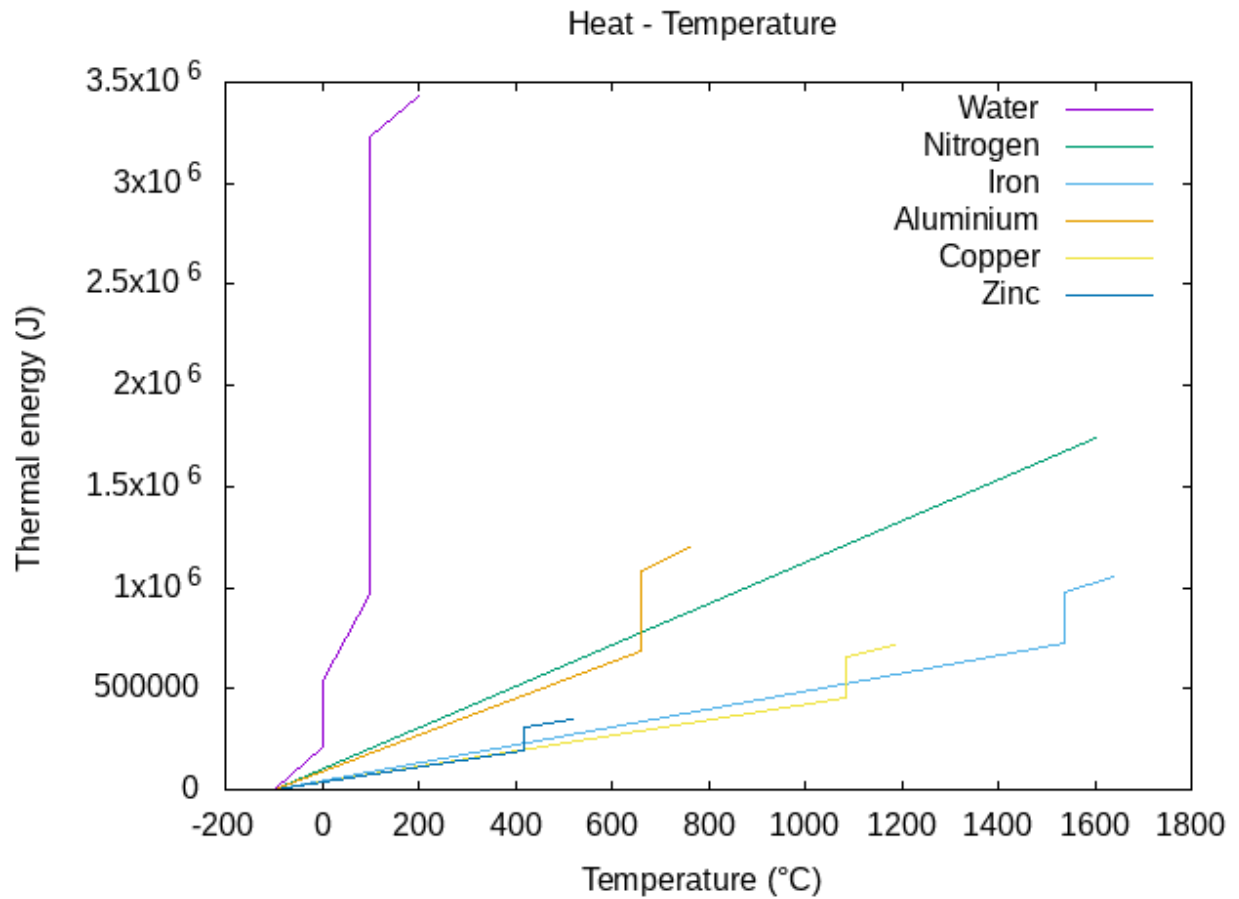
3.2.3 Energy for fusion



At 101325 Pa (normal atmospheric pressure)

| Material | Cp [solid] | Fusion Tc | Fusion heat | Cp [liq-uid] | Boiling Tc | Vaporization heat | Cp [gaz] |
|----------------|------------|-----------|-------------|--------------|------------|-------------------|----------|
| Units | J/(kg.K) | °C | J/kg | J/(kg.K) | °C | J/kg | J/(kg.K) |
| Water (H2O) | 2108 | 0 | 334000 | 4187 | 100 | 2264705 | 1996 |
| Nitrogen (N2) | | -210 | 25700 | | -196 | 200000 | 1025 |
| Iron (Fe) | 444 | 1538 | 247134 | 820 | 2861 | 6214000 | |
| Aluminium (Al) | 897 | 660.3 | 396590 | 1180 | 2519 | 10859000 | |
| Copper (Cu) | 385 | 1084.6 | 206170 | 490 | 2562 | 4721000 | |
| Zinc (Zn) | 380 | 419.5 | 112402 | | 907 | 1819000 | |





Energy to melt 1 kg of metal:

| | | | | | | | | | | | | |
|-------------|---|------|---|------|---|---------|---|---------|---|-----|-------|------|
| Water-steam | : | 100 | * | 4187 | + | 2264705 | = | 2683405 | J | <=> | 0.745 | kW/h |
| Iron | : | 1538 | * | 444 | + | 247134 | = | 930006 | J | <=> | 0.258 | kW/h |
| Aluminium | : | 669 | * | 897 | + | 396590 | = | 996683 | J | <=> | 0.276 | kW/h |
| Copper | : | 1084 | * | 385 | + | 206170 | = | 623510 | J | <=> | 0.173 | kW/h |
| Zinc | : | 419 | * | 380 | + | 112402 | = | 271622 | J | <=> | 0.075 | kW/h |

Reminder:

$$1\text{bar} = 100000\text{Pa}$$

$$1\text{Pa} = 1\text{Nm}^{-2} = 1\text{Jm}^{-3}$$

$$1\text{cal} = 4.1855\text{J}$$

$$1\text{mol} = 6.022 * 10^{23}$$

3.2.4 Auxiliary table

| Material | Molar mass | Specific volume | Thermal conductivity | Electrical resistivity | Magnetic permeability |
|----------------------------|------------|--------------------------|----------------------|-------------------------|-----------------------|
| Units | kg/mol | m ³ /kg | W/(m.K) | Ohm.m | H/m |
| Ice (H ₂ O) | 0.018 | 0.00109 | | | |
| Water (H ₂ O) | 0.018 | 0.00100 | 0.6 | 20 | 0.99 |
| Steam (H ₂ O) | 0.018 | 1.24 | | | |
| Nitrogen (N ₂) | 0.02802 | 0.799 | 0.026 | 10 ⁻⁹ | 1.00 |
| Iron (Fe) | 0.05584 | 127.0 * 10 ⁻⁶ | 80.4 | 9.70 * 10 ⁻⁸ | 5000 |
| Aluminium (Al) | 0.02698 | 370.4 * 10 ⁻⁶ | 237 | 2.65 * 10 ⁻⁸ | 1.00 |
| Copper (Cu) | 0.06354 | 111.6 * 10 ⁻⁶ | 384.1 | 1.68 * 10 ⁻⁸ | 0.99 |
| Zinc (Zn) | 0.06539 | 140.0 * 10 ⁻⁶ | 116 | 5.90 * 10 ⁻⁸ | 0.99 |

4.1 Alumina capacitor

4.1.1 Alumina

Alumina or *aluminium oxide* (Al_2O_3) seems to have good properties to be used as insulator to build a capacitor.

Aluminium oxide characteristics:

- density: 3.987 kg/dm^3
- thermal conductivity: $30 \text{ W.m}^{-1}\text{.K}^{-1}$
- dielectric strength: 14.6 MV.m^{-1}
- relative permittivity: 9.0

Insulator thickness for a *breakdown voltage* of 3.5V:

$$\frac{3.5}{14.6 * 10^6} = 0.24 * 10^{-6} \text{ m} = 0.24 \mu\text{m}$$

4.1.2 Capacitor geometry

We consider a a stack of *parallel-plate capacitor*.

- Insulator layer thickness: $0.24 \mu\text{m}$
- square surface of $0.1 \text{ m} \times 0.1 \text{ m} = 0.01 \text{ m}^2$
- height: 0.1 m
- volume: $0.1 * 0.1 * 0.1 = 1 \text{ dm}^3 = 1 \text{ L}$
- mass: $1.0 * 3.987 = 4 \text{ kg}$
- number of layers:

$$\frac{0.1}{0.24 * 10^{-6}} = 416000 \text{ layers}$$

4.1.3 Capacitance

For one layer:

$$C = \epsilon_0 * \epsilon_r * \frac{\text{surface}}{\text{thickness}} = 8.85 * 10^{-12} * 9.0 * \frac{0.01}{0.24 * 10^{-6}} = 3.31 * 10^{-6} F$$

For the complete capacitor:

$$416000 * 3.31 * 10^{-6} = 1.38 F$$

4.1.4 Energy

Stored energy at 3.0V:

$$E = \frac{1}{2} * C * u^2 = 0.5 * 1.38 * 3^2 = 6.2 J = 0.0017 Wh$$

4.1.5 Specific energy

$$\frac{6.2 J}{4.0 kg} = 1.55 J/kg = 0.00043 Wh/kg$$

4.1.6 Energy density

$$\frac{6.2 J}{1.0 L} = 6.2 J/L = 0.0017 Wh/L$$

4.2 Commercial ultracapacitors

4.2.1 Skeleton technologies

<https://www.skeletontech.com/>

According to their datasheet:

- Specific energy: 5.3 Wh/kg
- Energy density: 6.4 Wh/L

4.3 Conclusion

The naive approach with *alumina* insulator provides poor result compare to the current state of the art of the industry.

- Specific energy ratio:

$$\frac{5.3}{0.00043} = 12325$$

- Energy density ratio:

$$\frac{6.4}{0.0017} = 3764.7$$

CHAPTER 5

Indices and tables

- `genindex`
- `modindex`
- `search`